\[ \mu g \ Cl^-/ml = g \ of \ NaCl \times 10^3 \times \frac{35.453}{58.44} \quad \text{Eq. 26-1} \]
\[ \mu g \ Br^-/ml = g \ of \ NaBr \times 10^3 \times \frac{79.904}{102.90} \quad \text{Eq. 26-2} \]
\[ \mu g \ F^-/ml = g \ of \ NaF \times 10^3 \times \frac{18.998}{41.99} \quad \text{Eq. 26-3} \]